

# Chlorophyll-enriched chitosan-CuO-ZnO nanoparticles for antimicrobial activities

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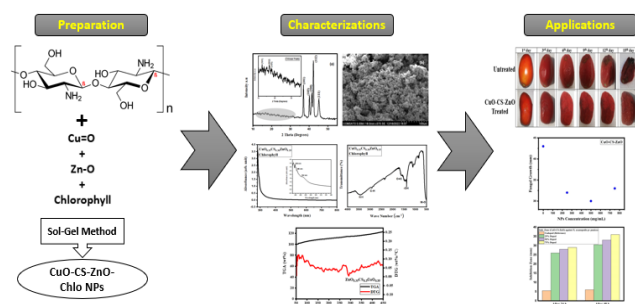
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Received: 11/05/2024, Accepted: 23/06/2024, Available online: 26/06/2024

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<https://doi.org/10.30955/gnj.006163>

## Graphical abstract



## Abstract

Chitosan-doped copper oxide and zinc oxide nanoparticles (CuO-CS-ZnO-Chlo NPs) were prepared utilizing a sol-gel synthesis route. Chlorophyll was added in the process of making NPs to enhance the effects of these NPs. The prepared NPs were thoroughly analyzed using advanced techniques including X-ray diffraction (XRD), scanning electron microscopy (SEM), UV-visible (UV-Vis) spectroscopy, Fourier-transform infrared (FTIR) spectroscopy, and thermogravimetric analysis (TGA). Structural analysis revealed the successful synthesis of irregularly shaped NPs with an average crystallite size of 10.7 nm. In optical properties, UV-Vis showed the absorption bands associated with different absorption centers in the sample and FTIR revealed various function groups associated with the preparation procedure and chemical used while synthesizing these NPs. On the application side, these NPs efficiently resist the growth of *X. axanopodis pv punicea* bacteria within 24-48 hours of investigation. The same NPs showed effective antifungal effects on the *Alternaria solani*. Finally, these NPs were used as food preservative agents whereas tomatoes were used as the target fruit. Tomatoes coated with NPs exhibited better preservation and a longer shelf life.

**Keywords:** Chitosan, chlorophyll, nanoparticles, antibacterial, antifungal, food preservation

## 1. Introduction

Nanotechnology unlocks new capabilities by utilizing the qualities of materials at a very small scale through the use of NPs (Ahila *et al.*; Benettayeb *et al.* 2023). It has been shown that nanomaterials can damage important components found within bacterial cells, therefore using nano-metal particles to fight germs can be an effective approach (Hermanto *et al.* 2024; Murtaza, Ahmed, *et al.* 2024; Said *et al.* 2024; Salas-Orozco *et al.* 2024). Nanoscale particles come in a variety of shapes and sizes, including atomic clusters, nanorods, dots, grains, fibers, films, and nanopores, all with large surface areas. These materials have better physicochemical properties than regular alternatives. They have been extensively explored and show promising antibacterial, antiviral, and antifungal effects (Nizami *et al.* 2021). Furthermore, NPs are known for their antibacterial characteristics, which inhibit the growth of microorganisms. This has resulted in its extensive use in dentistry. NPs are used in many dentistry sectors because of their potential to prevent infections and promote oral health (Dizaj 2015; Khurshid *et al.* 2015; Murtaza *et al.* 2024; Parnia *et al.* 2017). NPs, having distinct features, are used in the food industry for preservation. Their antibacterial activity, mediated by processes such as oxygen species and membrane disruption, improves food safety. Nanotechnology-based packaging provides a better barrier, mechanical strength, and biodegradability for longer food preservation (Ghosh *et al.* 2019). NPs offer intriguing possibilities for treating fungal infections by increasing medicine delivery. They improve medication properties, such as pharmacodynamics, resulting in lower toxicity and extended effects. Nanotechnology promises novel

methods for antifungal treatment and larger uses in infectious disorders (Renzi *et al.* 2021). Metal oxide NPs have strong antibacterial properties without affecting surrounding tissues, making them useful in textiles, water treatment, medicine, and food packaging. Their use reduces the toxicity concerns associated with organic disinfectants (Hajipour *et al.* 2012; Naseem and Durrani 2021; Singh *et al.* 2008; Tang and Zheng 2018).

CS, a versatile biopolymer, is made from chitin, a structural component present in the exoskeletons of crustaceans such as shrimp and crab. Chitin is transformed into CS via a process known as deacetylation, which alters its chemical structure and increases its solubility and usefulness. Due to its qualities, it is useful for tissue engineering, medication delivery, and wound healing, all of which hold promise for improvements in industrial and medical applications (Kou *et al.* 2021; Wu *et al.* 2023; T. Zhang *et al.* 2023). CS NPs, made from CS, are nanoscale particles with numerous applications. With improved qualities such as biocompatibility and specific distribution, they show potential in various industries, including medicine and agriculture (Fan *et al.* 2024; Fonseca *et al.* 2024; Jalal *et al.* 2023). The strong antibacterial qualities of CS NPs make them useful in the fight against infections and the creation of new antimicrobial agents for a range of applications (Benettayeb *et al.* 2023; Jha and Mayanovic 2023; Poznanski *et al.* 2023).

Similarly, Ag, Cu, CuO, and ZnO are examples of metallic materials that have been utilized to create antibacterial materials. Bacteria and viruses cell membranes are susceptible to the release of cations and reactive oxygen species caused by metallic materials. ZnO and CuO are two of the most commonly utilized metallic oxides because they are inexpensive, simple to produce, and have antibacterial qualities that make them useful in biological applications (Abdelrazek *et al.* 2023; Alturki 2022; Bolaina-Lorenzo *et al.* 2022; Boshkova *et al.* 2023; Govindasamy *et al.* 2023; Matyjasik *et al.* 2022). ZnO NPs improve food packaging films made of biopolymers by providing better fresh food preservation and minimizing zinc migration, which complies with safety standards (Chen *et al.* 2024; Espitia *et al.* 2012; Kim *et al.* 2022; Zare *et al.* 2022; W. Zhang *et al.* 2023). Fruit shelf life is extended when CuO NPs are incorporated into CS and hydroxypropyl cellulose-based bio-nanocomposites. This emphasizes the potential of these materials for active food packaging (Gunaki *et al.* 2024; Nemr *et al.* 2024; Saleem *et al.* 2024; Shankar *et al.* 2024). Orange juice shelf life is increased with polylactic acid sheets coated with polyaniline, ZnO, and CuO, which improves their antioxidant and antibacterial properties (Abdolsattari *et al.* 2022; Neethidevan *et al.* 2023; Serouti *et al.* 2024). Utilizing the qualities of CS and a variety of metal oxides, CS metal oxide NPs provide environmentally friendly methods of pollution removal for environmental remediation applications (Almaieli *et al.* 2022; Baroudi *et al.* 2023; Jiang *et al.* 2023; Rajivgandhi *et al.* 2023). CS-ZnO NPs provide a strong defense against microbiological

challenges in applications related to food safety, healthcare, and sanitation by combining the antibacterial qualities of CS with the enhanced activity of ZnO (Ibrahim *et al.* 2024; Salama and Aziz 2023). With their strong antibacterial, antibiofilm, antioxidant, and anticancer properties, CS-CuO NPs are proving to be a versatile agent in healthcare with the potential for a wide range of medical applications and therapeutic advances (Alturki 2022; Alvi *et al.* 2024; Sarfraz *et al.* 2023)

Chlorophyll, the green pigment found in plants and algae, plays a crucial part in photosynthesis, absorbing light energy for converting carbon dioxide and water into glucose and oxygen. Its bright color depicts the essence of biological life processes (Agathokleous *et al.* 2020; Ong *et al.* 2024; Zhang *et al.* 2024). Combining chlorophyll with nanoparticles improves drug delivery, imaging, and antibacterial efficiency, making it useful in biomedicine and food preservation. Mixing chlorophyll with CS improves its stability, effectiveness, and antioxidant action. CS is a carrier, preventing chlorophyll decomposition and increasing absorption, enhancing its potential health benefits (Alkahtani *et al.* 2020; Balusamy *et al.* 2022; Hanafy *et al.* 2021; Pucci *et al.* 2021). Mixing chlorophyll with CuO NPs enhances its antimicrobial properties due to the synergistic effect between chlorophyll and Cu ions. This mixture can successfully suppress microbial development, which may help with several biomedical and food preservation applications (Alves Batista de Souza *et al.* 2024; Gautam *et al.* 2016; Liu *et al.* 2022; Perreault *et al.* 2010; Sharma *et al.* 2019). When ZnO and chlorophyll merge, ZnO's antioxidant and UV-blocking properties are strengthened. This combination offers a strong defense against damaging UV rays and oxidative stress, making it a promising candidate for use in sunscreen and skin care products (Adil *et al.* 2022; Chen *et al.* 2023; Haghightazadeh 2021; Siddiqui *et al.* 2019).

A convincing aspect of the current investigation is the effect of chlorophyll-enriched CuO-CS-ZnO NPs on *X. axonopodis pv. Punicae* bacterial strain. *Xanthomonas axonopodis pv. punicae* primarily attacks pomegranate trees, leading to pomegranate bacterial blight. It can also infect other plants of the same family, including several species of the Punica genus and certain ornamental plants from the Lythraceae family. The study also investigates the implications of *Alternaria solani* growth, which is the causal agent of diseases affecting tomatoes, eggplants, and other vegetable crops. Furthermore, the study investigates the effect of CuO-CS-ZnO-Chlo NPs coatings on tomato preservation during subsequent storage. Furthermore, this study focuses on the optical, thermal, and morphological aspects of CuO-CS-ZnO-Chlo NPs.

## 2. Materials and methods

### 2.1 Material needed

Chitosan powder, acetic acid (CH<sub>3</sub>COOH, 99.5%), hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>, 98%), sodium hydroxide (NaOH, 97%), ammonia solution (NH<sub>3</sub>), ethanol (C<sub>2</sub>H<sub>5</sub>OH), copper acetate monohydrate (Cu (CH<sub>3</sub>COO)<sub>2</sub>·H<sub>2</sub>O, 99%), zinc

acetate dihydrate ( $\text{Zn}(\text{CH}_3\text{COO})_2 \cdot 2\text{H}_2\text{O}$ , 99%). Chemicals were purchased from Sigma-Aldrich. Additionally, spinach leaves were used for chlorophyll extraction.

## 2.2 Extraction of Chlorophyll from Spinach Leaves

Spinach leaves were crushed and mixed with ethanol. Later, the solution was filtered to get the chlorophyll.

## 2.3 Preparation of $\text{CuO}_{0.35}\text{CS}_{0.3}\text{ZnO}_{0.35}$ -Chlorophyll NPs

A 1.0% (w/v) CS solution was dissolved in 1% (v/v) acetic acid. 17.5 mL of 0.1 M Cu ( $\text{CH}_3\text{COO})_2 \cdot \text{H}_2\text{O}$ , 99% and 17.5 mL of 0.1 M Zn ( $\text{CH}_3\text{COO})_2 \cdot 2\text{H}_2\text{O}$  were added to 15 mL of CS solution. That solution was combined with the chlorophyll solution at a 1:1 ratio (v/v). To reach a pH of 7-10, the combinations were adjusted with 0.1 M NaOH and thoroughly mixed for 4 hours at 100 °C. Finally, 3 hours of drying at 120 °C and 4 hours of annealing at 450 °C.

## 2.4 Characterization and Instrumentation

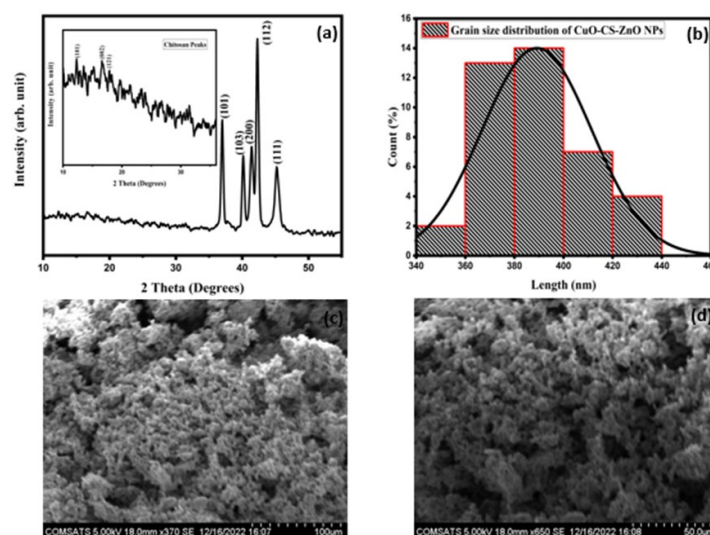
The crystal structure of CuO-Cs-ZnO-Chlo NPs was evaluated by XRD. Diffraction patterns from 10° to 55° were obtained at 50 kV and 40 mA. Cu-K $\alpha$  radiation (BTX-646) was used for the XRD evaluation. By utilizing the Shimadzu-1800 UV-Vis spectrometer, the absorbance of the produced NPs was measured between 280 and 800 nm. The Shimadzu FTIR-8400 model was used to perform FTIR spectroscopy across a 400–4000  $\text{cm}^{-1}$  range. SEM (SU-1500) was used to analyze the structure of NPs. Utilizing TGA (STA 449 F3 version), the thermal stability of CuO-CS-ZnO-Chlo NPs was evaluated. This was done in air-environment at temperatures ranging from 60-540 °C, with a heating rate of 10 °C/min and a flow rate of 50 mL/min. The antibacterial assessments focused on the Gram-negative bacterium *X. axanopodis pv punicea*. The antibacterial efficacy of the material was evaluated using the agar plate method (MIR-154-PE). Additionally, the agar well diffusion technique was used to study antifungal activities of the synthesized NPs on *Alternaria solani*. Tomatoes went through a series of steps to access food protection. They were first cleaned and sanitized with colloidal silver solution, then drained to remove any remaining water. The tomatoes were then treated by immersing them in water, letting them drain and air dry at room temperature. An Agrocolor colorimeter was used to measure changes in the surface color of the tomatoes. To understand the red component (R) and green component (G) using the CIE Lab scale [ $L^*$  (lightness),  $a^*$  (red-green color component)], a white teflon plate was calibrated. The texture properties of the tomatoes, including hardness and compression load, were assessed using a Brookfield CT3 texture meter in piercing and compression tests. HI 208 P.H. meter was used to determine the tomatoes' pH. Furthermore, a 503 nm wavelength was used to generate a lycopene calibration curve using a Genesys 10S UV-Vis spectrophotometer.

## 3. Results and discussion

### 3.1 Structural analysis

XRD analysis is critical for understanding the structure of the synthesized material and ensuring that the nanocomposites are successfully created. The XRD pattern

in Figure 1(a) shows a detailed structural composition with distinct crystal peaks and noticeable amorphous characteristics. The provided sample reveals trichotomous structures. It highlights the intricate essence of CS and illustrates its trichotomous character by embodying amorphous, crystalline, and transitional properties. The inset of Figure 1(a) distinctly illustrates XRD peaks located at 12.5°, 16.6°, and 17.8°, corresponding to the (101), (002), and (121) crystallographic planes, respectively. These crystallographic planes are characteristic of CS, with an orthorhombic structure and lattice parameters  $a = 8.2$ ,  $b = 16.4$ , and  $c = 10.3$  in concordance with the JCPDS card: 00-039-1894. The three discernible XRD peaks at 37°, 40°, and 42.5° are related to ZnO, which have been allocated to the (101), (103), and (112) crystallographic planes, respectively. These results confirm ZnO's hexagonal wurtzite structure, consistent with previous investigations (Zaman *et al.* 2022). The crystallographic planes (200) and (111) of CuO are represented by prominent XRD peaks at 41.5° and 45° respectively. These findings support previous studies that found that CuO has a cubic crystal structure (Magesh *et al.* 2018; Thakar *et al.* 2022). The results are also consistent with the information on JCPDS card 01-078-2076. The detection of three distinct phases in the XRD pattern signifies the accomplished synthesis of CuO-CS-ZnO-Chlo NPs. Utilizing the Scherrer formula, the crystallite size of the NPs was calculated (Collins *et al.* 2012; Kawsar *et al.* 2024). The average crystallite size of ~10.7 nm was found. SEM images of the NPs are depicted in figure 1(c&d). At a given magnification, it is clear that the particles are not separated. Particle clustering exhibiting clear agglomeration and further showing the existence of voids within the material. Figure 1(b) shows the grain size distribution of the prepared NPs. These results are consistent with earlier research (Alharbi *et al.* 2023; Benazir *et al.* 2011).

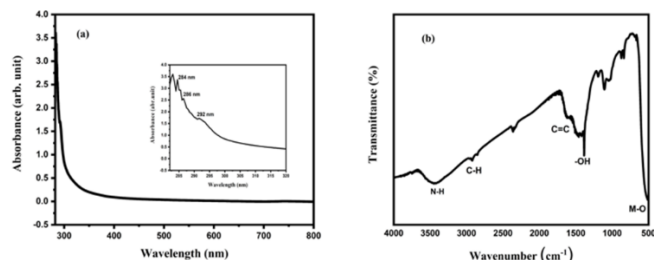


**Figure 1.** XRD pattern (a) size distribution (b) SEM image at 100  $\mu\text{m}$  (c), and SEM image at 50  $\mu\text{m}$  (d) of CuO-CS-ZnO-Chlo NPs

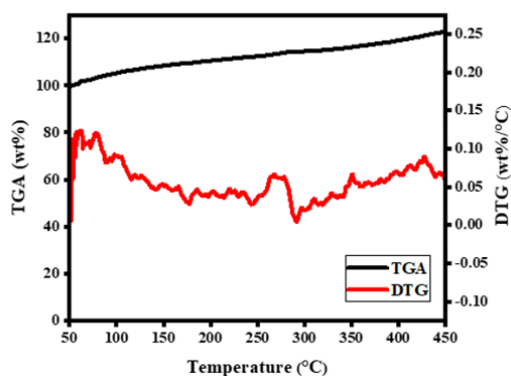
### 3.2 Optical analysis

Figure 2(a) shows the UV-Vis absorption spectrum of CuO-CS-ZnO-Chlo NPs. Sample absorption ceases around 450 nm. On zooming in the initial part of the absorption

spectrum, three different absorption bands can be seen. Initial strong absorption could be associated with the absorption of glass material used as the substrate for NPs. Additional absorption bands in the range between 283–300 nm are associated with CuO-CS-ZnO-Chlo NPs. (Alshammari 2022; Arab-Bafrani *et al.* 2021; Gupta *et al.* 2015; Keabadile *et al.* 2020; Khazaal *et al.* 2020; Muiz *et al.* 2022; Nandiyanto *et al.* 2019; Rilda *et al.* 2022). FTIR spectrum of CuO-CS-ZnO-Chlo NPs is depicted in figure 2 (b). Functional groups associated with the formation of NPs are observed in FTIR. The transmittance peak at 500  $\text{cm}^{-1}$  indicates the presence of metal ions, Cu or Zn, that are linked to the oxygen atoms. The presence of hydroxyl (-OH) groups is indicated at 1380  $\text{cm}^{-1}$ . This is compatible with the hydrophilic nature of metal oxides and the -OH groups in the chlorophyll molecule. The stretching vibrations of carbon-carbon double bonds (C=C) correspond to the band around 1650  $\text{cm}^{-1}$ . This signal shows the presence of unsaturated organic molecules, possibly chlorophyll with conjugated double bonds. The intense peak at 2920  $\text{cm}^{-1}$  corresponds to the stretching vibrations of carbon-hydrogen (C-H) bonds. This is typical of organic compounds and further suggests the presence of CS and possibly other organic components. The presence of amino (-NH) groups is shown by the broad and strong band around 3460  $\text{cm}^{-1}$ . This also suggests the presence of CS that includes amino groups (Asgari-Targhi *et al.* 2021; Kayani *et al.* 2015; Magesh *et al.* 2018; Rathore *et al.* 2020; Sankar *et al.* 2014; Seydi *et al.* 2019; Zhang *et al.* 2008).



**Figure 2.** UV-vis absorption curve. The inset shows the magnified curve (a) and FTIR pattern (b) of CuO-CS-ZnO-Chlo NPs with the inclusion of chlorophyll



**Figure 3.** TGA-DTG patterns of CuO-CS-ZnO-Chlo NPs

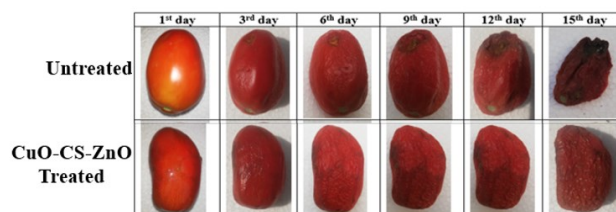
### 3.3 Thermal evaluation

Figure 3 shows how the TGA-DTG results profoundly reveal the temperature-dependent effects of CuO-CS-ZnO-Chlo NPs. These NPs show initial weight gain that an

oxidation process might have caused. Later, with small fluctuations, the weight of the materials stays constant throughout the measured temperature range. The material's enhanced thermal stability results from the synergistic co-doping of CS with ZnO and CuO. Minimal weight loss and modest changes imply thermal stability, the absence of volatile components, or effective decomposition resistance. Research shows strong bonding and high-purity materials exhibit such patterns, indicating their suitability for high-temperature applications. (Jayaramudu *et al.* 2019; Karpuraranjith and Thambidurai 2017; Kumar *et al.* 2019; Ltaief *et al.* 2021; Valiollahi *et al.* 2019; Wang *et al.* 2012).

### 3.4 Tomato preservation using CuO-CS-ZnO NPs

Tomatoes' color, which is important for market acceptance, varies as they ripen because of variations in their lycopene and chlorophyll contents. This change is quantified by parameters like  $L^*$  and  $a^*$ , where  $L^*$  stands for lightness and  $a^*$  for the ripeness indicator—the move from green to red. CuO-CS-ZnO-Chlo NPs-treated tomatoes postpone ripening-associated color changes by maintaining constant  $L^*$  levels (33–41) and less variable  $a^*$  values (31–44) over the course of a 15-day testing session. CuO-CS-ZnO-Chlo NPs prevent softening by preserving cell wall structure and turgor pressure, which also helps to retain firmness. Taste is impacted by the pH rising from 3.5 to 4 while being stored. Tomatoes that have been treated exhibit longer color retention and a reduced lycopene concentration over time. Through the modulation of membrane features during refrigeration at 12–16  $^{\circ}\text{C}$ , NPs have a good impact on quality parameters, contributing to practical preservation and increasing shelf life (Anthon and Barrett 2012; Belović *et al.* 2015; Del Real-López *et al.*; Esserti *et al.* 2024; Gopinath *et al.* 2024; Murtaza, Rizwan, *et al.* 2024; Murtaza, Usman, Iqbal, Tahir, *et al.* 2024; Pinheiro *et al.* 2013; Saei *et al.* 2011).

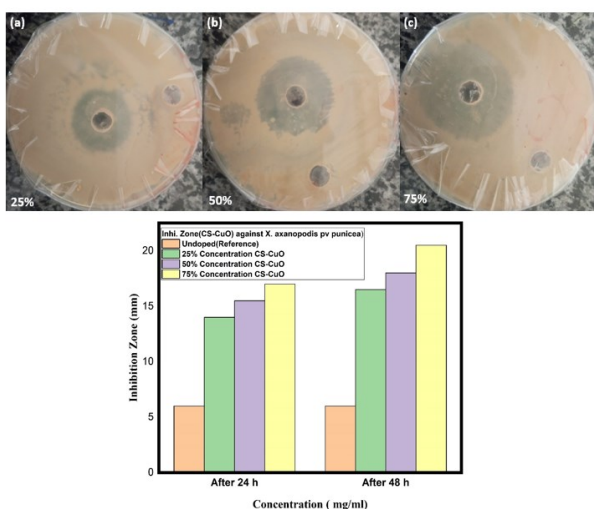


**Figure 4.** Preservation of tomatoes by utilizing CuO-CS-ZnO-Chlo NPs

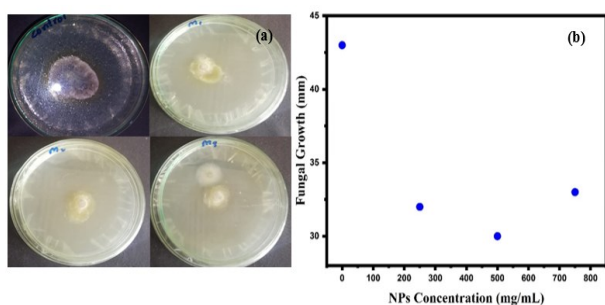
### 3.5 Antibacterial effects

Figures 5 (a, b, c & d) display how well CuO-CS-ZnO-Chlo NPs stopped the growth of harmful bacteria. When more NPs were used, they worked better against *X. axanopodis pv punicea* bacteria after 24 and 48 hours. These NPs showed significant antibacterial effects and hence could be used to cure seeds and leaves suffering from bacterial diseases. In previous investigations, these NPs were found to exhibit substantial antibacterial activities against a wide range of diseases, demonstrating their efficiency in limiting microbial development. The NPs have demonstrated exceptional stability and biocompatibility, making them appropriate for a variety of applications,

including medical and agricultural industries. After 24 hours the inhibition zone was 26 mm for 25%, 28 mm for 50%, and 29 mm for 75% of NPs. Later, the same observation was done after 48 hours and the results were: 30 mm for 25%, 33 mm for 50%, and 35 mm for 75% of NPs in addition, chlorophyll also played a possible role in the enhancement of antibacterial effects. The treated seeds and plants exhibited reduced infection symptoms, highlighting the importance of these NPs in protecting against harmful bacteria, and contributing to healthier crops (Sibiya and Sumbwanyambe 2019). The interaction between nano-Zn and nano-Cu and bacteria is initiated by Cu and Zn adhesion to the cell membrane. This results in morphological changes, membrane depolarization, and intracellular leakage, which ultimately leads to cell death (Fayaz *et al.* 2010; Roy *et al.* 2019).



**Figure 5.** Antibacterial activity of CuO-CS-ZnO-Chlo NPs with the inclusion of chlorophyll against *X. axanopodis pv punicea*. Discs showed (a) 25% (b) 50% (c) 75% of NPs concentrations and (d) a bar chart depicting the inhibition zone after 24 and 48 hours at three different concentrations



**Figure 6.** (a) Discs used for determining antifungal activity at different NPs concentration (b) antifungal activity of CuO-CS-ZnO-Chlo NPs against *Alternaria solani*

**3.6 Antifungal effects**

CuO-CS-ZnO-Chlo NPs were used in this study as a biocontrol agent against the *Alternaria solani* fungus. Significant antifungal efficacy was demonstrated by agar well diffusion analysis, shown in figure 6 (a), which showed a discernible decrease in *Alternaria solani* growth with increasing CuO-CS-ZnO-Chlo NPs concentrations. The computed observed growth rates at 0 mg/mL, 250 mg/mL, 500 mg/mL, and 750 mg/mL concentrations of CuO-CS-

ZnO-Chlo NPs are shown in figure 6 (b). The corresponding measurements are 43 mm, 32 mm, 30 mm, and 33 mm. This demonstrates the promising effect of CuO-CS-ZnO-Chlo NPs on fighting the targeted fungus, especially when combined with chlorophyll. These NPs were very effective in inhibiting the fungus, even at low concentrations and beneficial for the health of plants (Mohamed A. Mosa and Sozan E. El-Abeid 2023; M. A. Mosa and S. E. El-Abeid 2023)

**4. Conclusions**

CuO-CS-ZnO NPs containing chlorophyll were synthesized by utilizing the sol-gel method. Structural and optical analysis confirmed the effective synthesis of NPs with an average crystallite size of 10.7 nm. Thermal analysis revealed the stability of the prepared CuO-CS-ZnO-Chlo NPs. CuO-CS-ZnO-Chlo NPs showed good antibacterial activity against *X. axanopodis pv punicea* bacteria and modest antifungal properties for *Alternaria solani*. Elevated NP concentrations boosted both antibacterial and antifungal activities. The use of these NPs has the potential to improve preservation quality, particularly by extending the shelf life of tomatoes.

**Declaration of competing interest**

The authors declare that they have no competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

**Acknowledgments**

We extend our sincere gratitude to Dr. Muhammad Farooq Qayyum and Dr. Muhammad Sajid from the Department of Soil Science, Bahauddin Zakariya University, for their invaluable assistance with X-ray crystallography (XRD). The authors also extend their appreciation to the Researchers supporting project number (RSP2024R470), King Saud University, Riyadh, Saudi Arabia.

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