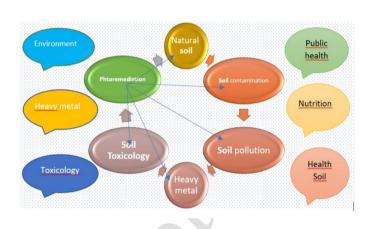
- 1 Use of phytoremediation for pollution removal of hexavalent chromium-
- 2 contaminated acid agricultural soils
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14 GRAPHICAL ABSTRACT



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17 Abstract

Chromium is a common heavy metal pollutant found in industrial wastewaters which may 18 19 pollute agricultural soils through groundwater and watering. Phytoremediation is an economical 20 and highly applicable method for removal of pollutants from agricultural soils. This research was carried out for the removal of hexavalent chromium (Cr (VI)) contamination from the soil with 21 the phytoremediation method. For this purpose, only 30 mg kg⁻¹ hexavalent chromium (Cr (VI) 22 as Chromium CrO₃, only 10 mL bacteria Rhodobacter capsulatus DSM1710 and chromium plus 23 24 bacteria applied to the pots and Malabar spinach (Basella alba L.) grown in the pots. At the end 25 of experiment the results showed that side branching, leaf width, plant dry weights were the 26 highest agro-morphological traits when bacteria were applied to chromium polluted soil. Some 27 macro and micro nutrient elements which are essential for plant nutrition were analyzed (N, P, 28 K, Ca, Mg, Fe, Cu, Mn and Zn). Among them, N, P, Fe, Cu, Mn and Zn were found to be

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statistically significant at the level of 5%. The Cr content of Malabar spinach in control soil was 29 0.31mgkg⁻¹, but it was 2.33mgkg⁻¹ when the soil was contaminated with Cr at the end of 30 experiment. Moreover, when bacteria were additionally applied the Cr content increased to 4.02 31 mgkg⁻¹ of Malabar spinach. Chromium pollution antagonistically affected both some nutrient 32 33 element (P, K, Ca; Mg) and some heavy metals (Fe, Cu, Zn, Mn) in the soil. This study shows that phytoremediation can be used to remove the soil pollution caused by containing high 34 35 hexavalent chromium. For this reason, the nitrogen fixing bacterium Rhodobacter capsulatus and the hyperaccumulator Malabar spinach plant can be used. It is the first study where Malabar 36 spinach was used a hyperaccumulator plant for chromium pollution in the soils. 37

Keywords: Toxicity, Phytoremediation, macro and micro-elements, Cr (VI), *Rhodobacter capsulatus*

40 **1. Introduction**

41 Continuously changing and evolving technology affect agriculture directly. Various 42 applications have been carried out in agriculture such as chemical fertilizers, hormones, soil 43 regulators, pesticides, sludge and wastewater usage for watering in order to obtain the highest 44 efficiency from unit area. On the other hand, fast and unbalanced increase in population, 45 urbanization and industrialization cause environmental problems. Among these problems resides 46 the heavy metal pollution of the soil and water sources (Adiloğlu 2016; Adiloğlu et al., 2016). 47 Metals such as Ag, Cd, Cr, Cu, Hg, Mn, Mo, Ni, Pb and Zn are soil polluters which can be 48 uptake by the plants from the soil. Chromium is one of the common heavy metals that pollute the 49 agricultural soils as a result of industrial activities. It is a transition element in VI-B group with proven toxic effect. It can take various values ranging from $^{+1}$ to $^{+6}$. Chromium is found as 50 chromium oxide in the soil and has two oxidation states: Cr⁺³ and Cr⁺⁶ (Bebek 2001; Türsün 51 52 2017) and Cr^{+6} is more toxic than Cr^{+3} . The total chromium level allowed in the agricultural soils is 100 mg/kg, while the extractable level is only 1 mg/kg (Adiloglu 2013). Generally, chromium 53 54 in the soil varies between 7-760 mg/kg depending on the source (Demir 2008). Some plants are reported to be highly efficient in terms of heavy metal remediation from the soil and they are 55 56 tolerant to heavy metal toxicity. Using plants for removal of pollutants from the soil is called phytoremediation which can be applied easier than physicochemical technologies. Advantages of 57 58 phytoremediation are being effective both on organic and inorganic pollutants and lower 59 expenses of system set up and amelioration. The system can be used in both natural and artificial 60 environments and the dimensions of the contaminated area are not disadvantageous in 61 phytoremediation using plants. Phytoremediation using plants is a cheap option in these 62 circumstances (Sadowsky 1999; Yinanç and Adiloğlu, 2017). Being economical is indeed one of 63 the important advantages of phytoremediation. The cost of cleaning of a Pb contaminated area 64 was calculated to be 6 times lower by phytoextraction than engraving the soil over 30 years of 65 time (Cunningham, 1996). The economic value of phytoremediation for farmers was calculated to be around 15,000 € over a period of 20 years (Lewandowski et al. 2006). The primary factor 66 67 affecting the success of phytoremediation is the heavy metals in the soil becoming available for uptake by the plant roots. For this reason, complex forming chelates are used which increase the 68 69 uptake of metals by plants. There are various phytoremediation methods depending on the plant 70 to be used and the pollutant. The choice of the method therefore depends on the uptake and 71 removal mechanisms of the plants, chemical and physical properties of the pollutant, the 72 suitability of the phytoremediation method to the pollutant, the concentration and the depth of 73 the pollutant in the soil and climate (EPA 2004). Malabar spinach belongs to the Basellaceae 74 family (Deshmukh and Gaikwad, 2014). There are two taxonomic varieties: Basella rubra L. ve 75 Basella alba L. They are differentiated by their leaf properties and stem color (Adhikari et al., 76 2012; Cook 2010; Deshmukh and Gaikwad, 2014; Ray and Roy, 2007). The origin of Malabar 77 spinach is in India and Indonesia and can be naturally grown in tropical Asia (Saroj et al., 2012). It is known that green plants respond to nitrogen containing inorganic fertilization, but nitrogen 78 79 application can affect yield up to a certain point. Nitrogen plays an important role in vegetative 80 development and yield quality. Increased nitrogen application can positively or negatively affect 81 some agro-morphological properties, macro and micro nutrition contents and quality of the 82 product. (Lemaire and Gastal, 2009). Rhodobacter capsulatus is a photosynthetic Gram negative 83 purple non-sulfur bacterium (PNSB) that lives in soil and fresh water. It possesses nitrogenase 84 enzyme through which it can fix free nitrogen from the air. This bacterium has been subject of 85 many researches due to its versatile metabolism, hydrogen production and nitrogen fixation 86 abilities (Weaver et al., 1975). It can utilize many organic substances therefore can be used for 87 wastewater and sewage treatment and bottom mud contaminated with organics. (Nagadomi et 88 al., 2000; Sasaki et al., 1998). It is known that heavy metal contamination decreases the bacteria 89 amount in soil and this affects the soil vitality (Ding et al., 2017). R. capsulatus and other some 90 PNSB have been shown to be used for bioremediation of certain heavy metal contaminations in 91 soil (Ge et al., 2017; Kis et al., 2015). Moreover, R. capsulatus was shown to reduce hexavalent 92 chromium to less toxic trivalent chromium (Merugu et al., 2013; Rajyalaxmi et al., 2017). 93 Bahadur et al. (2017) and Polti et al. (2011) studied different bacteria (rhizobacteria and 94 Streptomyces sp. MC1) for the removal of chromium pollution from the soils. According to the 95 researchers, different bacteria positively affected and decreased the chromium pollution in the 96 soils. The aim of this study is using phytoremediation, a cheap and efficient biological method 97 for cleaning soil contaminated with chromium of low mobility. It was revealed that the Malabar

98 spinach *Basella alba* L. is a hyperaccumulator plant together with the applied bacterium R. 99 *capsulatus*. To the best of our knowledge, this study is the first example of employing R. 100 *capsulatus* to enhance the hexavalent chromium phytoremediation capacity of the Malabar 101 spinach.

102 **2. Materials and methods**

103 2.1. Setting up and running the experiment

104 This study was carried out in pots under controlled conditions in November 2017-March 2018 in Namık Kemal University Faculty of Agriculture Soil Science and Plant Nutrition 105 Department (40°98'N, 27°48'E). The study was designed in triplicates according to randomized 106 107 block design. A total of 12 pots were used in the study. For each trial, 3 pots were used: 3 control 108 pot, 3 pots for only bacteria application, 3 pots for chromium application, and 3 pots for the 109 application of chromium and bacteria together. The day temperature was around 27°C and night 110 temperature was 21-22°C with humidity not less than 85%. The plants obtained light for twelve 111 hours per day during the experiment. No pesticides were used during the experiment. The 112 standard variety of Malabar spinach (Zengarden Firm) was used for the research. Seeds were sown in multi-celled trays filled with peat (Klasmann- Deilmann, Potground H, Germany) in 113 114 November 2017. Chromium (VI) oxide (CrO₃) (Sigma-Aldrich No: 232653) was used as the heavy metal. It was dissolved in distilled water to have the concentration of 30mgkg⁻¹ CrO₃. 115 116 Later this water was applied to the pot soil 30 days before transplantation of the plants from the 117 multi-celled trays. After the germination and generation of 2-3 leaves (19 days after 118 germination), the plants were transplanted to pots of 4 kg as 1 plant/pot.

119 Rhodobacter capsulatus DSM1710 was grown in modified Biebl Phennig medium (Biebl 120 and Pfennig, 1981) containing 20 mM acetate and 10 mM glutamate as carbon and nitrogen 121 sources, respectively. Bacteria were grown anaerobically in fully filled glass bottles under 122 constant illumination with 2000 lux light intensity at 30°C. In the mid exponential phase of 123 growth, bacteria were collected by centrifugation at 4100 rpm for 20 min. The bacterial pellet 124 was washed twice with sterile saline solution. The pellet was re-suspended in sterile distilled 125 water to reach a final concentration of 10⁷ CFU/mL and 10 mL bacteria were applied to the root area of the plants in pots. The applications in this study were as follows: Control, only bacteria, 126 only chromium, chromium + bacteria. The peat used for germination contained 160-260 mgl⁻¹ N, 127 $180-280 \text{ mgL}^1 \text{ P}_2\text{O}_5$, 200-150 mgL⁻¹ K₂O, and 80-150 mgL⁻¹ Mg. The pH of the turf was 6. The 128 129 organic matter of the peat was 70% and the C content was 35%.

130 2.2. Soil and plant analysis

The soil used in the study contained 3.9% organic matter, and 5.2% lime. The EC $x10^6$ of the 131 132 soil was 700. The changeable potassium (K₂O) in the soil was 128 kg da⁻¹, while available phosphorous (P₂O₅) amount was 9.25 kg da⁻¹. The pH value of soil samples was determined in 133 134 1:2.5 soil: water using a pH meter, the lime contents of soil samples were determined by a 135 calcimeter, organic contents were determined by Smith-Weldon method, and phosphorus contents were determined by NaHCO₃ method (Sağlam, 2012). The salt contents of soil samples 136 137 were determined with EC meter (U.S. Soil Survey Staff, 1951). The texture of soil samples was 138 evaluated according to Bouyoucos method (Bouyoucos 1955; Tuncay, 1994). The available Zn, 139 Cu, Fe, Mn and extractable chromium contents of the soil samples were analysed with ICP-OES using a buffer solution (DTPA method:0.005 M DTPA+0.01M CaCl2+0.1 M TEA (pH:7.3)) 140 141 (Lindsay and Norvell, 1978). The pH of the soil, CaCO₃ content, electrical conductivity, organic matter content, available P, exchangeable K, available Zn, Fe, Mn, Cu, and extractable Cr and 142 143 texture given in Table 1 (Bouyoucos 1955; Jackson 1967; Kacar 1995; Lindsay and Norvell, 144 1978; Olsen and Sommers, 1982; Sağlam 2012).

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 Table 1. Some physical and chemical properties of the experimental soil

Physical and chemical properties	Values	Reference			
_pH (1:2.5)	6.43	Sağlam, 2012			
EC (µs/cm)	1533	U.S. Soil Survey Staff, 1951			
CaCO ₃ (%)	5.84	Sağlam, 2012			
Organic matter (%)	1.91	Sağlam, 2012			
Texture	Clay loam	Bouyoucos 1955; Tuncay, 1994			
Available P (mgkg ⁻¹)	41.39	Olsen and Sommers, 1982			
Exchangeable K (mgkg ⁻¹)	262.75	Kacar, 1995			
Available Mn (mgkg ⁻¹)	0.81	Lindsay and Norvell, 1978			
Available Cu (mgkg ⁻¹)	1.79	Lindsay and Norvell, 1978			
Available Fe (mgkg ⁻¹)	0.37	Lindsay and Norvell, 1978			
Available Zn (mgkg ⁻¹)	0.79	Lindsay and Norvell, 1978			
Extractable Cr (mgkg ⁻¹)	0.45	Lindsay and Norvell, 1978			

Plants were harvested after 120 days after germination. Plant height, number of plants, leaf width, leaf length, number of side branches, wet and dry weights of the Malabar spinach were measured. Some macro and micro nutrition element contents (P, K, Ca, Mg, Fe, Cu, Mn, Zn, and Cr) of plants were determined with ICP-OES (Agilent 700 series) after wet decomposition and N content was determined with Kjeldahl method in Namık Kemal University Central Research Laboratory (Kacar and Inal, 2010; EPA 1996).

152 2.3 Statistical analysis

Statistical analysis of the results was carried out with Analysis of variance (ANOVA) and
Duncan's Multiple Range Test using Statistical Package for Social Sciences (SPSS) Version 21
(IBM 2012).

156 **3. Results and discussion**

157 The images from experimental setup and Malabar spinach during experiment and after the harvest can be seen in Fig. 1. The plants in the pots at the start of the experiment (Fig. 1. a) 158 and at the end of the experiment (Fig. 1. b) were photographed. The Malabar spinach of the 159 160 control condition can be seen in Fig. 1. c, while the comparison of the control with a plant of 161 bacterial application was given in Fig. 1. d (left: control, right: bacterial application). The 162 improvement of plant growth by addition of R. capsulatus is quite noticeable; the bacterial 163 addition has stimulated plant growth in this experiment. The effects of chromium and bacteria 164 applications on the Malabar spinach were given in Table 2.



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Figure. 1. Images of experimental process and harvest of the Malabar spinach. a) Start of the experiment, plants in pots. b) Plants on the day of harvesting. c) Overall image of Malabar spinach (control experiment pot). d) Malabar spinach of control (left) and bacteria application (right).

- 170
- 171 **Table 2.** Effects of chromium and bacteria treatments on agro-morphological traits of Malabar

		spinach					
Treatment s	Plant height (cm)	Number of leaves (unit)	Leaf height (cm)	Leaf width (cm)	Number of Side branch (unit)	Wet weight (g)	Dry weight (g)
Control	148±36.3 ^{ns}	40±9.5 ^{ns}	12±0.0 ^{ns}	8.6 ± 0.2^{b}	10.6±4.0 ^a	44±11.4 ^{ns}	$4.94{\pm}0.0^{a}$
Bacteria	185±27.5 ^{ns}	31±4.6 ^{ns}	12±0.7 ^{ns}	9.1±0.6 ^{ab}	7.0±1.5 ^a	47±5.6 ^{ns}	3.11±0.0°
Chrome (Cr ⁺⁶)	149±16.8 ^{ns}	30±4.0 ^{ns}	11±0.6 ^{ns}	8.4±0.4 ^b	2.0±1.0 ^b	35±6.0 ^{ns}	$3.58{\pm}0.0^{b}$
Chrome and bacteria	135±5.0 ^{ns}	33±6.3 ^{ns}	12±0.8 ^{ns}	10.3±1.0 ^a	5.6±0.8 ^{ab}	45±4.8 ^{ns}	3.52±0.0 ^b

173 All the values are mean \pm standard error (SE), n=3. Different letters (a, b, c) indicate significances at p≤0.05, ns: 174 non-significant

175 Although there have been increase and decrease in plant height, number of leaves, leaf 176 height and wet weight of the Malabar spinach by applications of bacteria and chromium alone 177 and together, they were found to be statistically insignificant. However, changes in leaf width, 178 number of side branches and dry weight were statistically significant. The negative effects of 179 heavy metal application were obvious on these biological traits when compared to the control 180 condition. The addition of *R. capsulatus* bacteria in heavy metal contamination showed the most 181 significant effect on the width of Malabar spinach leaves. The decrease in the leaf widths by 182 heavy metal application were reversed by bacterial activity.

183 **Table 3.** Effects of chromium and bacteria treatments on macro nutrition elements content of

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Treatments	N (%)	P (%)	K (%)	Ca (%)	Mg (%)
Control	4.57 ± 0.4^{c}	0.47 ± 0.02^{b}	2.87±1.22 ^{ns}	1.47±0.17 ^{ns}	0.92±0.05 ^{ns}
Bacteria	$5.07{\pm}0.3^{b}$	0.58±0.01 ^a	5.03±0.05 ^{ns}	1.87±0.04 ^{ns}	1.04±0.01 ^{ns}
Chromium (Cr ⁺⁶)	5.56±0.7 ^a	0.37±0.02°	2.51±1.08 ^{ns}	1.66±0.24 ^{ns}	0.88±0.07 ^{ns}
Chromium and bacteria	5.15±0.17 ^b	0.43±0.01 ^b	4.03±0.05 ^{ns}	1.69±0.04 ^{ns}	0.96±0.02 ^{ns}

185 All the values are mean \pm standard error (SE), n=3. Different letters (a, b, c) indicate significances at p≤0.05, ns:

186 non-significant, each element was evaluated individually

187

The effects of heavy metal and bacteria applications on macro nutrient elements of 188 189 Malabar spinach were given in Table 3. The nitrogen content of the Malabar spinach increased 190 with application of heavy metal compared to the control condition. The highest nitrogen content 191 was obtained when heavy metal was applied alone. The reason may be a synergistic effect 192 between chromium and nitrogen. Moreover, there is an increase in nitrogen content in heavy 193 metal and bacteria application together. Higher acquisition of nitrogen from the soil was 194 suggested to be a mechanism of stress avoidance (Blaudez et al., 2000). On the other hand, the 195 lowest nitrogen content amount all the conditions were observed in bacteria only application. 196 The reason can be that bacteria may also utilize the nitrogen in the soil. *Rhodobacter capsulatus* 197 has many different metabolisms. It can fix nitrogen form the soil, but can also shift its 198 metabolism and can consume the nitrogen available in the soil for growth and maintanance. The 199 experiment period was 2 months, and for better evaluation of nitrogen contents, the experiment

200	duration should be increased and field researches should be conducted. As the pot experiments
201	took only two months, in order to see the maximum bacterial utilization process, the experiments
202	should continue in greenhouse and field. Nitrogen plays an important role in biological
203	properties, yield and quality of the plants. Nitrogen deficiency negatively affects the vegetative
204	development of the plant as nitrogen is a crucial element for green parts of the plants (Lemaire
205	and Gastal, 2009; Özer 2003). In case of deficiency, leaf and stem structures would be weak and
206	vegetative development period would be short (Güneş et al., 2007; Karaman et al., 2012; Smith
207	and Cassel, 1991). In a previous study done with komatsuna, as the nitrogen was applied as 10
208	kg da ⁻¹ , its effect on plant length, dry and wet weights was higher than other doses of no
209	nitrogen, 15 g da ⁻¹ and 20 kg da ⁻¹ nitrogen (Acikgoz et al., 2014). This showed that as well as
210	deficiency, over application of nitrogen can have negative effects. Phosphorus content was the
211	highest under bacteria only application. Phosphorus is an element known for its generative
212	developmental effect on the plants. Besides, phosphorus negatively affects plant vegetative
213	growth (Adiloğlu et al., 2011; Güneş et al., 2007; Karaman et al., 2012). Potassium, calcium and
214	magnesium contents were not significantly affected by heavy metal and bacteria applications.
215	Potassium amount in plants has an impact on resistance against diseases and pests. In case of
216	potassium deficiency, the opening and closing metabolism of the stoma are disrupted. This
217	increases the chance of bacterial and fungal infections in the plant (Öktüren Asri and Sönmez,
218	2005). Calcium is vital for plant development and cell wall synthesis as 90% of the calcium take
219	place in the cell wall. Magnesium has an active role in energy metabolism in the roots (Karaman
220	et al., 2012; Mikkelson 2010). The chromium heavy metal contamination did not negatively
221	affect the amounts of these elements, which enhanced the phytoremediation capacity of the
222	Malabar spinach.

223 **Table 4.** Effects of chromium and bacteria treatments on micro nutrition elements content of

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Malabar spinach shoot

(Malabar spinach shoot				
Treatments	Fe (mgkg ⁻¹)	Cu (mgkg ⁻¹)	Mn (mgkg ⁻¹)	$Zn \ (mgkg^{-1})$	Cr(mgkg ⁻¹)
Control	82.70±2.01 ^a	$9.63 {\pm} 0.26^{b}$	116.60±5.69 ^b	32.17±1.52 ^{ab}	0.31±0.10 ^c
Bacteria	81.57±1.73 ^b	9.90±0.06 ^b	164.97±2.81ª	36.53±0.52 ^a	0.39±0.03°
Chromium (Cr ⁺⁶)	43.57±2.92°	8.13±0.17°	126.63±9.70 ^b	31.20±2.0 ^b	2.33±0.03 ^b
Chromium and bacteria	59.27±0.79 ^{bc}	11.43±0.09 ^a	162.47±3.38 ^a	36.13±0.47 ^a	4.02±0.0.48 ^a

225 All the values are mean \pm standard error (SE), n=3. Different letters (a, b, c) indicate significances at p ≤ 0.05 ,

226 each element was evaluated individually

227 An antagonistic effect between Fe and Cu contents was observed in the chromium heavy metal 228 applied pots. This situation is obvious from the statistically different groups in the analysis 229 results. The change in the contents of Mn and Zn were found to be nonsignificant. The contents 230 of Fe and Mn were found to be statistically different at 5% level in the pots with bacteria 231 application. When evaluated with increased N contents in soil upon bacteria application, the 232 nitrogenase of this bacterial species can be suggested to be active in this study. This enzyme 233 contains Fe in the structure. Therefore, it may be suggested that Fe was consumed by bacteria, 234 and there may be synergistic effect between Fe and Mn. In the pots with both chromium and 235 bacteria application it was observed that bacteria could compensate the negative effects of 236 chromium and also positively affected Cu, Mn and Zn plant nutrient elements. The dual 237 application of chromium and bacteria also increased the accumulation of chromium in the plant.

The Malabar spinach was shown to accumulate the heavy metal chromium (Table 4). The chromium content in the control was 0.31 mgkg⁻¹, this was the chromium content of the Malabar spinach accumulated from the soil without any additional heavy metal application. The addition of the bacteria to the soil, again without additional heavy metal, did not significantly enhance chromium uptake of the Malabar spinach from the soil.

243 However, when 30 mgkg⁻¹ chromium was applied to the soil, Malabar spinach could accumulate 2.33 mgkg⁻¹ chromium according to the control. This shows that Malabar spinach 244 245 can act as a heavy metal hyper accumulator plant. Its accumulation of the heavy metal chromium was even enhanced with the addition of the bacteria to the soil and increased to 4.02 mgkg⁻¹. 246 247 This proves that application of *R. capsulatus* increased the phytoremediation of chromium from 248 the soil by the Malabar spinach. Moreover, Cu, Mn and Zn, but not Fe contents were improved 249 compared to the control when chromium was applied together with the bacteria. Chromium and 250 bacteria have antagonist effect on Fe but synergistic effect on Cu, Mn and Zn contents of 251 Malabar spinach. Similar results were obtained earlier researches with sunflower (Helianthus 252 annuus L.), spiny chicory (Cichorium spinosum), black gram (Vigna mungo) different plants (Antoniadis et al., 2017; Bahadur et al., 2017; Saravanan et al., 2019). 253

- **Table 5.** Effects of chromium and bacteria treatments on macro nutrition elements content of
- 255

soils

Treatments	N (%)	P (%)	K (%)	Ca (%)	Mg (%)
Control	0.30 b	0.015 a	0.548 a	0.301 a	0.247 a
Bacteria	0.68 a	0.012 b	0.486 b	0.278 c	0.194 b
Chromium (Cr ⁺⁶)	0.69 a	0.013 ab	0.496 b	0.271 d	0.172 c

Chromium and	0.68 a	0.015 a	0.538 a	0 291 h	0.205 b
bacteria	0.00 a		0.330 a	0.291 b	0.205 0

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All the values are mean \pm standard error (SE), n=3. Different letters (a, b, c) indicate significances at p \leq 0.05, each element was evaluated individually

258

Nitrogen content of the soil increased according to control with bacteria, bacteria and chromium and only chromium applications (Table 5). Available phosphorus content of the soil increased with only bacteria application. But exchangeable K content of the soil decreased with only bacteria and chromium application. Exchangeable Ca and Mg changed in the same way. These increases and decreases were found to be statistically significant at the level of 5 %. These values were obtained after the experiment (Table 5).

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- 266

267 **Table 6.** Effects of chromium and bacteria treatments on micro nutrition elements content of

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		soils		
Treatments	Fe (mgkg ⁻¹)	Cu (mgkg ⁻¹)	Mn (mgkg ⁻¹)	Zn (mgkg ⁻¹)
Control	2.69 a	0.22 ab	6.953 b	0.47 a
Bacteria	2.40 c	0.21 bc	6.233 d	0.47 a
Chromium (Cr ⁺⁶)	2.36 c	0.20 c	6.406 c	0.45 b
Chromium and bacteria	2.59 b	0.24 a	9.086 a	0.45 b

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All the values are mean \pm standard error (SE), n=3. Different letters (a, b, c) indicate significances at $p \le 0.05$, each element was evaluated individually

271 Generally, available Fe, Cu, Mn and Zn contents of the soil decreased with bacteria and 272 chromium applications according to control (Table 6). Iron nutrient element were lower in all 273 274 the trials compared to the control pots most probably due to the consumption of Fe by bacteria. 275 Because of the antagonist relationship between chromium and copper, the lowest Cu was 276 observed in the pot with only chromium application. This decreased was shown to be 277 compensated by the bacterial presence. All the micro nutrients except Zn decreased in the only 278 bacterial application. The reason may be the uptake and use of these elements by the additional 279 bacteria. However, Zn content did not significantly change but it has a synergistic relationship 280 with chromium. But these element values decreased with bacteria plus chromium applications 281 except Mn contents. These increases and decreases were found to be statistically significant at the level of 5%. Chromium application (30 mgkg⁻¹) decreased Fe, Cu, Zn and Mn contents of the 282 283 soil. Chromium pollution negatively affected some micro nutrient element contents in the soil. But the negative effects of chromium were decreased with bacteria plus chromium application.

All heavy metal values were determined after the harvest of the plants (Table 6).

286 **4. Conclusions**

287 Industrial and agricultural activities result in pollution of water and soils which are 288 important environmental parameters. Heavy metal pollution is one of the leading causes of water 289 and soil pollution. It was shown that phytoremediation can be easily and economically applied to 290 accumulate chromium from the soil which was polluted with chromium contaminated soil. The 291 nitrogen fixing R. capsulatus, which belongs to an important group of bacteria for soil biological 292 activity and plant nutrition, increased phytoremediation capacity of the Malabar spinach. 293 Different treatments in this study significantly affected the nitrogen content positively, 294 phosphorous content negatively; and other macro nutrient elements were not significantly affected in Malabar spinach shoot. Similarly, among the micro nutrient elements, Fe content was 295 296 negatively affected while Cu, Mn and Zn contents were positively affected by heavy metal and 297 bacteria treatments in Malabar spinach. It can be seen that the applied *R. capsulatus* bacterium is 298 especially effective on acquisition of these elements.

299 The results of the study revealed that Malabar spinach can be used in phytoremediation of 300 heavy metal contaminated soil together with soil application of the bacterium Rhodobacter 301 capsulatus. Yet, field experiments should be carried out for more certain inference about their 302 use in large scale. However, the results may indicate that R. capsulatus may be a resistant 303 bacterium for chromium contamination in soil, and can be added to the soil contaminated with 304 chromium for remediation purposes. This study suggests that soil contaminated with chromium cleaned by Malabar spinach, and R. capsulatus was shown to increase the phytoremediation 305 306 efficiency of Malabar spinach. In order to have an information on R. capsulatus growth in the 307 soil, R. capsulatus counting in a soil sample can be conducted in the future. Moreover, a detailed 308 analysis on how the interaction of addition of this bacterium to soil with other bacteria can be 309 done by a high throughput microbiome study.

The Malabar plant has been used for the first time in the literature to remove chromium contamination from soils. This plant can be a hyper accumulator for chromium. Malabar spinach a hyper accumulator plant in such cases may take up the heavy metals from the soils, and hence clean the soils polluted with some heavy metal contaminated soils.

314

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