

1 **Analysis of Spatial-temporal and Ion Characteristics Change of Precipitation**
2 **in the Southwest of China, from Policy Perspective**

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4 Luo Y.H.¹, Wu L.J.², Xiao Y.N.², He R.J.^{3*}

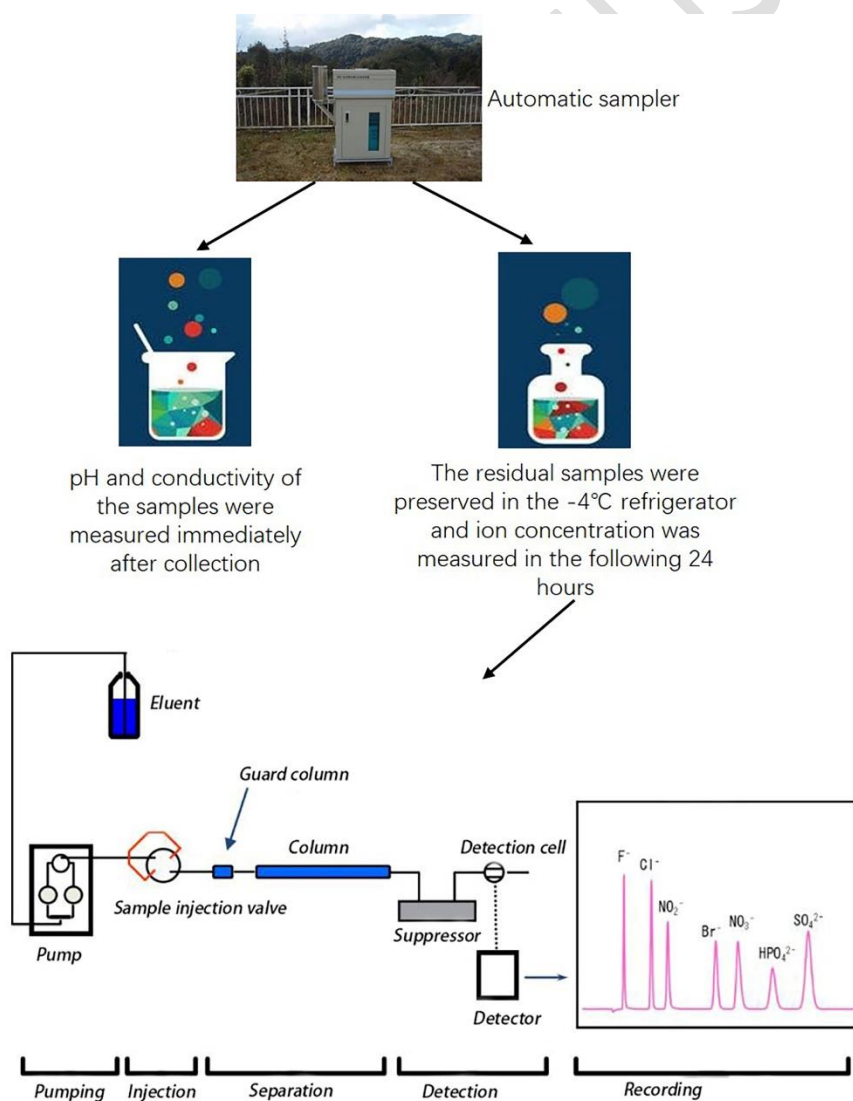
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6 ¹ School of basic medical, Southwest Medical University, Luzhou 646000, China

7 ² Luzhou Environmental Monitoring Center, Luzhou 646000, China

8 ³ Department of Environmental and Industrial Hygiene, College of Public Health, Southwest
9 Medical University, Luzhou, 646000, Sichuan, China

10 *Corresponding author: herenjiang@163.com

11
12 **GRAPHICAL ABSTRACT**



13

14

15 **Abstract:** According to the data from Luzhou environmental monitoring center from 2015 to 2018,
16 the tendency of spatial-temporal and ion characteristics in precipitation have been analyzed in order
17 to demonstrate the precipitation pollution status. Traditionally, the environment may become worse
18 as the development of the economy, especially in the developing countries. However, the results
19 show that the quality of precipitation has been improved during these four years. The annual
20 average pH value of precipitation in Luzhou City increased, and the frequency and conductivity of
21 acid rain declined, which indicates that the quality of precipitation improved annually. This may due
22 to a strictly political strategy enacting. The decrease of the equivalent concentration ratio of SO_4^{2-}
23 and NO_3^- demonstrated the transformation of pollution type, evolving from a typical sulfuric acid to
24 a mixed type of sulfuric and nitric acid, indicating the economic transformation. The correlation
25 coefficient between SO_4^{2-} and NO_3^- was high. This may be because SO_2 and NO_2 emitted from
26 industrial enterprises entered the atmosphere together in the same way. The strong correlation
27 between F^- and SO_4^{2-} , NO_3^- may relate the pollution sources such as cement and glass enterprises. It
28 is necessarily to strengthen the management of relative enterprises.

29

30 **Keywords:** pH, precipitation, ion characteristics

31

32 1. Introduction

33

34 Air pollution has been a universal concern in many countries because of its harmful impacts on
35 public health, ecosystem, and economy. The substances of air pollution contain sulfur dioxide (SO_2),
36 nitrogen oxides (NO_x), excessive carbon dioxide (CO_2), Volatile Organic Compounds (VOCs),
37 particulates (PM_{2.5}, PM₁₀), etc.(Siddiqi and Farsi 2019). These emissions are released to the
38 atmosphere either by natural processes or human activities (US EPA, 2017a). Air pollutants lead
39 numerous of environmental problems such as acid rain, climate changes, crop and forest damage
40 and adverse impacts on wildlife. The effect of this pollution costs approximately \$5 trillion per year
41 to cure people and repair the environmental damage (World Bank, 2016). Therefore, more works
42 along with governments and environmental agencies to decrease air pollution and damage it causes
43 are needed.

44 One the environmental consequences of air pollution is atmospheric acid deposition including
45 dry deposition and wet deposition. Precipitation is often recognized as wet deposition(Grimm and
46 Lynch 2004). Its acidity can be expressed as the magnitude of solution acidity (H^+ concentration or
47 pH). When the pH of precipitation is lower than 5.6, the precipitation is often renamed acid
48 rain/precipitation which is one of the global environmental problems (Galloway et al. 1976). The
49 performance of pH and other chemical parameters of precipitation has been monitored for many
50 years to ensure the quality of precipitation. From such monitoring activities, data are evident for
51 determining temporal and spatial deposition trend, predicting ecological effects, modeling
52 atmospheric processes, or planning future emission control strategies (Wisniewski and Kinsman
53 1988).

54 The significant chemical factors in acid rain are nitrate and sulfate because NO_x and SO_2
55 exhausted from mobile and stationary sources mainly lead to a decrease of the rain acidity by
56 chemical reactions (Wakida et al. 2001). Lightning and volcanic eruptions react with water
57 molecules or get mixed with dust are typical natural processes to form these two chemical
58 compounds. These molecules terminally fall as dry or wet deposition(Mutahharah et al. 2014). NO_x
59 and SO_2 as precursors of acid rain may cause secondary pollutants like particles and nitrogen
60 species(Menz and Seip 2004).

61 Acid rain predominates many countries such as the Canada, United States, China, and
62 Europe(Bowman 1992). Although many researchers analyzed the characteristics of acid rain all
63 over the world, there is limited literature addressing acid rain in China, especially in these highly
64 developmental economic periods. This essay analyzed the data of the spatial-temporal and ion
65 characteristics of precipitation in the recent four years in Luzhou city. The acidity of precipitation in
66 this city was extremely serious. We hope this article would help to provide useful information about
67 the quantity variation of ion characteristics and evaluate the quality of the atmosphere in Luzhou
68 city.

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70 **2. Sampling site**

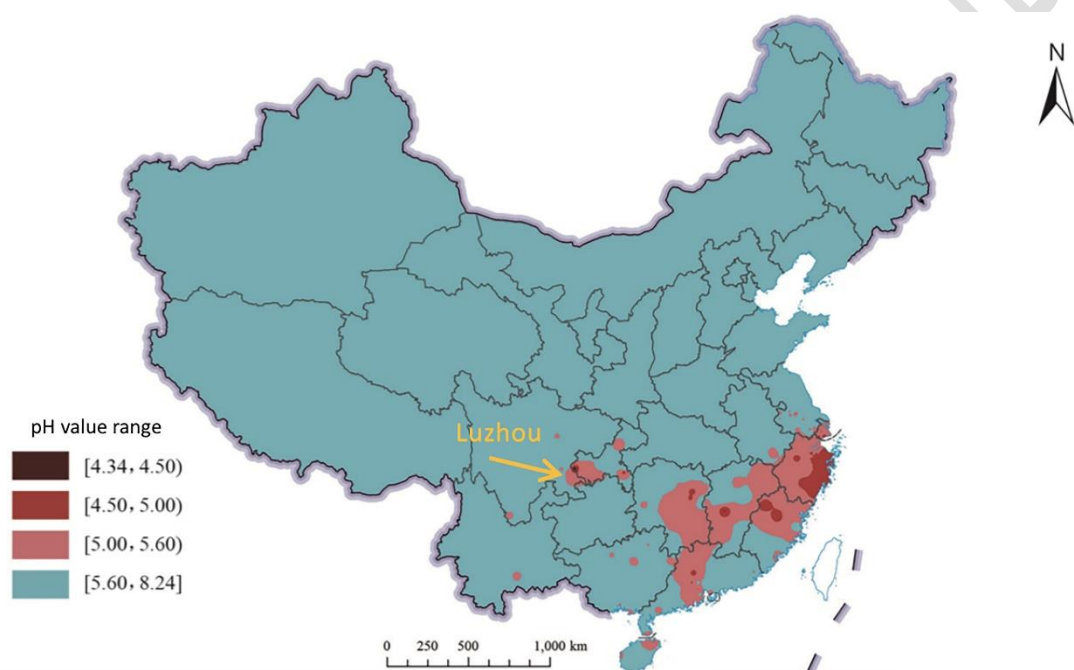
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72 Luzhou, N28°52'28.94", E105°26'20.32", is situated in the southwest of China.
73 Geomorphologically, it is dominated by medium mountains (1000-1902m above the sea level) and
74 low mountains (500-1000m above sea level). There are four distinct seasons in Luzhou because of

75 its typical continental monsoon climate. According to Luzhou people's government data, the
76 temperature and rainfall are relatively higher in Luzhou compared with surrounding cities. Rainfall
77 mainly focused from July to August and the least concentrated from December to March next year.

78 By the end of 2017, Luzhou Economic and Social Development Statistic Bulletin presented that
79 the population of the city has exceeded 5 million, and Gross Domestic Product was 159.6 billion
80 RMB, and the contribution of the three industries to the growth of the economy was 5.0%, 59.7%,
81 and 35.3% respectively. The second and third industries are still pillar industries.

82



83

84 Fig 1 The location of Luzhou City and the distribution of acid rain in 2018 in China (according to
85 Environmental Protection Administration of China)

86

87 3. Methods

88

89 The samples were collected with 500ml polyethylene bottles. They had been cleaned with
90 hydrogen chloride (2-3 mol/L) and rinsed with Milli-Q water (18.25MΩ) before use (Hua and Han
91 2011). The bottles were located in the automatic sampler that immediately collected the sample
92 when it was rain. The samplers were situated on the roofs 1.5m high above the ground, gathering 24
93 hours wet precipitation from 8 a.m. to 8 a.m. the next day as one sample to analyze, and 247
94 samples were collected.

95 We analyzed pH and conductivity of the samples immediately after collection. Then the
96 residual samples were preserved in the -4°C refrigerator and ion concentration was measured in the
97 following 24 hours after sampling. $0.45\mu\text{m}$ organic Millipore membrane was used to filtrate
98 samples and then immersed by deionized water for 24 hours (Shi et al. 2013). This membrane is an
99 inert material with features of Uniform aperture, fast rate passing and avoiding from the absorption
100 with other chemical factors in the sample. Therefore, the samples were able to prevent loss and
101 contamination of components to be tested.

102 Major anions (Cl^- , F^- , SO_4^{2-} , NO_3^-) were demonstrated by an ionic chromatography
103 (ICS-1500, 09090451). The detection limits of F^- , Cl^- , NO_3^- , SO_4^{2-} were 0.03, 0.03, 0.1 and 0.1
104 mg/L. Major cations (K^+ , Na^+ , Mg^{2+} , Ca^{2+} , NH_4^+) were measured by ionic chromatography (ICS-90,
105 7100483). The detection limits of K^+ , Na^+ , Mg^{2+} , Ca^{2+} , NH_4^+ were 0.01, 0.01, 0.01, 0.03 and 0.01
106 mg/L. Reagent and procedural blanks were measured in parallel to the sample treatment and the
107 blanks were below the detection limit of the measured species. The relative standard deviations
108 (RSD%) of the method were below 5%. Quantitative recoveries of spiked samples ranged from
109 85% to 120%.

111 4. Results and discussion

113 4.1 pH value, frequency of acid rain and conductivity

115 From the previous studies, CO_2 , NO_x , and SO_2 existing in the nature can dissolve into the
116 clouds and droplets, causing pH values of the rain in the clean atmosphere to be between 5.0-5.6
117 (Charlson and Rodhe 1982), (Galloway et al. 1993). Precipitation with pH below 5.0 results from
118 the presence of natural or anthropogenic emission of H_2SO_4 and/or HNO_3 , while that of above 6.0
119 might consider certain inputs of alkaline species into the precipitation. It is shown in Table 1. 247
120 rain samples were collected from 2015 to 2018. The average pH value observed in Luzhou city
121 increased from 4.28 in 2015 to 5.00 in 2018. Total rainfall is above 1000mm. The acid rainfall did
122 not show significant difference from 2015 to 2017, however, it was observed obvious decrease in
123 2108 ($P < 0.05$). Both maximum and minimum pH values of precipitation were observed in 2016.

124 Frequency of acid rain represents the intensity of acidity of precipitation. It equals the ratio of
125 the number of acid rains to the total number of rains, It is an important indicator to ensure whether

126 the certain area is an acid rain area besides pH value (Siddiqi and Farsi 2019). The frequency of
127 acid rain observed a distinct decrease from 86% to 46% during the four years with consistency to
128 the trend of pH value.

129 The conductivity of precipitation is considered to be a significant parameter in acid rain
130 chemistry monitoring. The conductivity of ions in precipitation is generally used in the quality
131 assessment on the analysis data of ionic components. Table.1 shows that the conductivity increased
132 slightly in the first two years (5.78 to 6.20 mS/m), then decreased obviously from 2016 to 2018
133 (6.20 to 3.67).

134 pH value calculating equation is as follows. n indicates the number of rains. V_i indicates the i -th
135 volume of rain.

$$136 \quad pH = -\log \frac{\sum_{i=1}^n 10^{-pH} \times V_i}{\sum_{i=1}^n V_i}$$

137 Table 1. The statistical results of precipitation in Luzhou, 2015-2018

Years	Number of samples	pH mean (in unit)	Total Rainfall (mm)	Acid rainfall (mm)	Frequency of acid rain	Max (in unit)	Min (in unit)	Conductivity (ms/m)
2015	51	4.28	1091.70	853.50	86%	6.64	3.25	5.78
2016	59	4.43	1022.80	917.20	85%	7.74	3.20	6.20
2017	79	4.41	1084.40	877.00	69%	6.74	3.55	4.67
2018	58	5.00	1005.30	390.10	46%	6.69	3.52	3.67

138
139 Therefore, pH value, frequency of acid rain and conductivity all present that the quality of
140 atmospheric precipitation is improving annually. SO₂ and NO₂ are the main acid-producing
141 precursors of precipitation (gao et al. 2001), (LIN et al. 1999). The main reason would be some
142 relative policies enacted in this area. The government urged the enterprises with large pollutant
143 emissions to enact strategy of “Total pollutant emission reduction” five years ago. According to the
144 statistical data from Luzhou Environmental Protection of Agency, SO₂ emissions decreased from
145 68,900 tons in 2014 to 36,900 tons in 2018. Meanwhile, NO₂ emissions declined from 38,200 tons
146 in 2014 to 24,400 tons in 2018.

147
148 *4.2 ionic composition*

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150 The equivalence ratio of the sum of anions to that of cations (Σ anions/ Σ cations) is usually
151 regarded as an indicator of the completeness of the measured major constituents (Al-Khashman
152 2005). In this study, the mean equivalent sum of anions of that cations were between 0.92 and 0.96,
153 which suggested that all major ions were measured.

154 Table 2 shows that, in these four years, the concentration of each cation was all in the order of
155 $\text{NH}_4^+ > \text{Ca}^{2+} > \text{Mg}^{2+} > \text{Na}^+ > \text{K}^+$. The proportion of NH_4^+ and Ca^{2+} mass concentration ranged from
156 78.6% to 90.8% of the total amount of cations, indicating that the main cations of precipitation was
157 NH_4^+ and Ca^{2+} . The order of mass concentration of each anion from high to low was: $\text{SO}_4^{2-} > \text{NO}_3^- >$
158 $\text{Cl}^- > \text{F}^-$, wherein the percentage of SO_4^{2-} and NO_3^- mass concentration ranged from 92.1% to 95.4%
159 of the total anion, showing that SO_4^{2-} and NO_3^- was the main anions of precipitation in Luzhou city.

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Table 2. The ion concentration of precipitation in each year in Luzhou, mg/L

Years	SO_4^{2-}	NO_3^-	NH_4^+	Ca^{2+}	Mg^{2+}	Cl^-	F^-	K^+	Na^+
2015	9.86	4.52	1.29	1.61	0.18	0.48	0.21	0.32	0.29
2016	9.32	4.38	2.12	2.77	0.18	0.83	0.34	0.42	0.46
2017	8.24	4.57	2.42	1.29	0.10	0.39	0.23	0.19	0.09
2018	5.64	3.39	2.36	1.24	0.13	0.49	0.19	0.22	0.13

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163 **4.3 Ions proportion change in precipitation**

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165 According to previous literature, precipitation is mainly dominated by sulfuric acid-type
166 precipitation in economically underdeveloped areas. With the rapid economic development in the
167 region, precipitation acidification will change from the previous sulfuric acid type to the mixed type
168 of sulfuric acid and nitric acid (Mai et al. 2010). It is shown in Fig.2 that the $\text{SO}_4^{2-}/\text{NO}_3^-$ equivalent
169 concentration ratio witnessed a decrease from the highest value of 4.4 in 2015 to 3.3 in 2018. It
170 represents that the pollution type of acid precipitation in Luzhou city has gradually evolved from a
171 typical sulfuric acid type to complex pollution of sulfuric acid plus nitric acid.

172

173 NH_4^+ and Ca^{2+} are the predominant neutralization substances in rainwater, however, the
neutralization by Mg^{2+} and K^+ is negligible (Hua and Han 2011). Fig.1 shows that the equivalent

174 concentration ratio of main cation $\text{Ca}^{2+}/\text{NH}_4^+$, which had a neutral effect in precipitation, increased
175 in the first two years, then declined from a peak of 2.5 in 2015 to 1.0 in 2018. The concentration of
176 NH_4^+ increased year by year shows that the content of NH_3 in the atmosphere of Luzhou has an
177 increasing influence on the quality of precipitation. Moreover, the only large-scale power plant
178 around the main city completed the ammonia desulfurization technical reform project in 2015. The
179 chemical Plant installed a selective catalytic reduction unit for denitrification treatment.

180 Generally, agricultural sources are regarded as the main source of NH_3 emissions in the
181 atmosphere (Kanakidou et al. 1995), (Hoek 1998), (Aneja and P. 2003). Besides, related research
182 has found that in recent years, motor vehicle exhaust and industrial emissions have become the
183 main non-agricultural sources of NH_3 in the atmosphere of urban areas (Whitehead et al. 2007),
184 (Vollenweider 2010). The number of cars in Luzhou city increased from 373,400 in 2015 to
185 568,200 in 2018, increasing by 52.2%. Also, the Chuan Nan Power Plant, a famous coal-fired
186 power plant around the main city, completed the ammonia desulfurization technical reform project
187 in 2015. Some Chemical Nitrification Plant in the city installed a selective catalytic reduction unit
188 for denitrification treatment. Chuan Tian Hua, a famous large-scale chemical enterprise producing
189 nitrocellulose, installed ammonia desulfurization in 2015 to treat SO_2 emissions from the enterprise.

190 These treatment facilities will inevitably have an “ammonia escape” situation during use. This
191 may be one of the reasons why the percentage of NH_4^+ equivalent in precipitation in Luzhou City
192 has increased year by year. However, the NH_4^+ equivalent concentration ratio decreased slightly
193 from 2017 to 2018. It may be because from 2015, Luzhou City has increased the management of
194 poultry breeding, delineated strictly banned areas and limited areas in the urban area and shut down
195 some livestock and poultry farms that do not meet the management requirements, resulting in a
196 decrease in NH_3 emissions from agricultural sources and a decrease in the percentage of NH_4^+
197 equivalents in precipitation.

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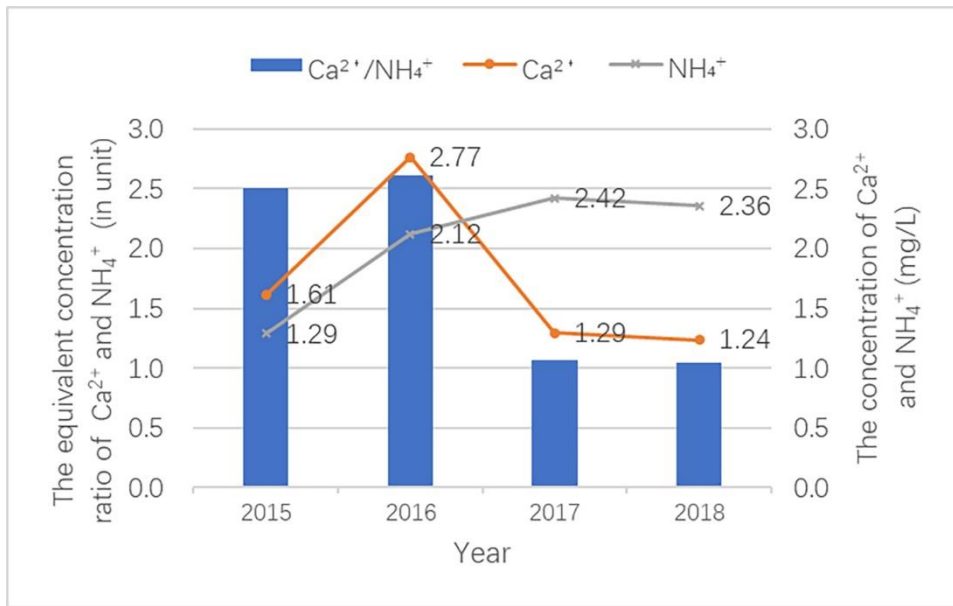


Fig 1. the equivalent concentration of $\text{Ca}^{2+}/\text{NH}_4^+$ and trend of Ca^{2+} and NH_4^+ concentration during 2015-2018

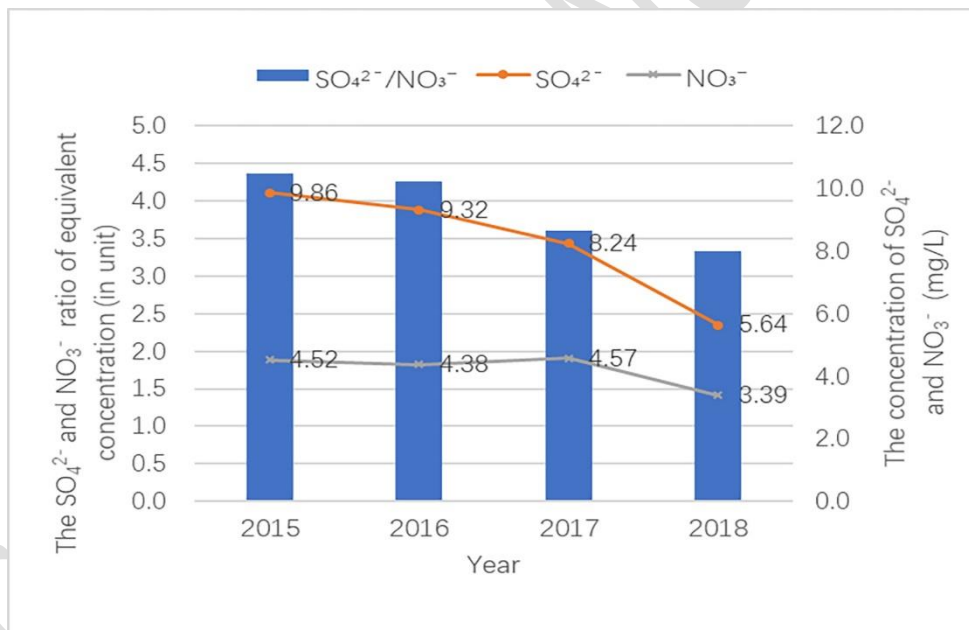


Fig 2. the equivalent concentration of $\text{SO}_4^{2-}/\text{NO}_3^-$ and trend of SO_4^{2-} and NO_3^- concentration during 2015-2018

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206 4.4 Trend analysis of characteristic ions in precipitation

207

208 An in-depth analysis of the correlation between the various ion components in the precipitation
 209 can show the relationship between the ions, and it is of great benefit to determine the source of the
 210 pollutants, to understand and master the influencing factors of acid rain formation and to effectively

211 control them (Nam et al. 2001). Pearson analysis was carried out on the correlation of ions in
212 precipitation in Luzhou City in 2018 via SPSS software. T-test was used to ensure the significance
213 test. The results are shown in the following table 3.

214 Table 3 The correlation analysis of ions in Luzhou precipitation in 2018

		SO ₄ ²⁻	NO ₃ ⁻	F ⁻	Cl ⁻	NH ₄ ⁺	Ca ²⁺	Mg ²⁺	Na ⁺	K ⁺
SO ₄ ²⁻	Pearson	1.000								
NO ₃ ⁻	Pearson	0.933	1.000							
F ⁻	Pearson	0.958	0.902	1.000						
Cl ⁻	Pearson	0.474	0.420	0.462	1.000					
NH ₄ ⁺	Pearson	0.971	0.911	0.913	0.568	1.000				
Ca ²⁺	Pearson	0.890	0.818	0.860	0.407	0.867	1.000			
Mg ²⁺	Pearson	0.772	0.712	0.758	0.425	0.809	0.853	1.000		
Na ⁺	Pearson	0.176	0.128	0.158	0.914	0.304	0.160	0.283	1.000	
K ⁺	Pearson	0.670	0.512	0.663	0.841	0.748	0.595	0.684	0.700	1.000

215 Note: the correlation analysis significance level is 0.01.

216 It can be seen from the data in the Table 3 that the correlation coefficient of NH₄⁺ with SO₄²⁻,
217 NO₃⁻, and F⁻ is higher in 2018, 0.971, 0.911, and 0.913, respectively. The correlation coefficient
218 between SO₄²⁻ and NO₃⁻ is as high as 0.933, which is a strong correlation between them. This may
219 be because, during the year 2018, SO₂ and NO₂ emitted by industrial enterprises entered the
220 atmosphere together in the same way. There is also a strong correlation between F⁻ and SO₄²⁻, NO₃⁻,
221 which is 0.958 and 0.902, respectively. This may be related to the pollution sources such as cement
222 and glass enterprises.

223

224 5. Conclusion

225

226 Based on the analysis of acid precipitation in recent four years, we can note that the Luzhou
227 government emphasizes the control of pollutant emission even though Luzhou City is in the process
228 of high economic development. This is beneficial to society and public health. The quality
229 improvement of acid precipitation also presented that the economic transformation of Luzhou City.

230

231 **Conflicts of interest**

232

233 The authors declare no conflict of interest.

234

235 **Acknowledgments**

236

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238 Department of Education.

239

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